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SHORT ANSWER Answer the 12. 18.

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nd edition) 1. Terminologies: solution,

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solute, solvent, miscible (vs. immiscible) liquids, dissolution, 2. Know the common types of solutions and give examples of each (Table 12.1) 3. Concepts: Be able to explain the following: (a) Intermolecular forces (IMFs) of attractions that exist in solution and relative

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Chapter 12 Review Solutions Section

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12 REVIEW Solutions SECTION 3 SHORT
ANSWER Answer the following questions
in the space provided. 1. Describe the
errors made by the following students in
making molar solutions. a. James needs

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a 0.600 M solution of KCl. He measures out 0.600 g of KCl and adds 1 L of water to the solid.

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Go to first question Chapter 12 Review Problems INSTRUCTIONS: You do not need to write the question, ONLY

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WRITE THE PROBLEM NUMBER and ANSWERS/SOLUTIONS. • For problems that involve calculations, you must show your work to get full credit. • For multiple choice questions, you can simply write the letter (a, b, c, or d) of the correct response.

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Chapter 12: Solutions ...

Answers to Chapter 12 Review Questions

1. Precambrian history is more difficult to decipher because of the lack of fossils, which hinders correlation of rocks. Also, rocks of this great age are metamorphosed and deformed, extensively eroded, and obscured by younger overlying strata. 2.

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(30) define & identify suspensions &
solutions define an alloy define &
identify electrolytes & nonelectrolytes
factors that affect the rate of dissolution
definitions of unsaturated, saturated &
supersaturated solutions general rules

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for predicting whether a solute is soluble in a solvent definition of solubility effects of temperature & pressure on the solubility of gases and solids in liquids definitions of molarity (M) and molality (m) solving ...

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Chapter 12 and 13 Review

Worksheet—ANSWERS OGT Section

Page Person, Place, Date, Term

Description 12.3-13.3 424, 449 Charles

Lindbergh Made the first nonstop solo

flight across the Atlantic Ocean in his

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plane the Spirit of St. Louis. He became a hero upon his return home, receiving a ticker tap parade and a reception from the white house.

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3 SHORT ANSWER Answer the following questions in the space provided. 1. Describe the errors made by the following students in making molar solutions. a. James needs a 0.600 M solution of KCl. He measures out 0.600 g of KCl and adds 1 L of water to the solid.

CHEM 142 REVIEW GUIDE- CHAPTER

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12: SOLUTIONS

This Chapter 12 Review, Section 3: Solutions Worksheet is suitable for 9th - 12th Grade. A neat presentation and challenging content make this solutions worksheet an ideal homework assignment for your budding chemists. Short answer and problem solving questions get them to review molarity,

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