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the energy-releasing process by which a substance changes from a gas or vapor to a solid without first becoming a liquid. Diffusion the movement of one material through another from an area of higher concentration to an area of lower concentration.

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particles occupy a negligible volume; particles exert minimal forces on one another; particles are in constant random motion and collisions are perfectly elastic (KE of the system does not change) YOU MIGHT ALSO LIKE...

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the process by which a substance changes from a gas or vapor to a solid without becoming a liquid Boyle's law states the volume of a fixed amount of a gas held at a constant temp. varies inversely with pressure

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Name Date Class 76 Chemistry: Matter and Change • Chapter 13 Study Guide for Content Mastery. Section 13.3 Liquids and Solids. In your textbook, read about liquids and solids. In the space at the left, write true if the statement is true; if the statement is false, change the italicized word or phrase to make it true.

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