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An aqueous solution of lead(II) chloride is mixed with an aqueous solution of sodium sulfide to form an aqueous solution of sodium chloride and a lead(II) sulfide precipitate. b. When gaseous sulfur trioxide is passed into liquid water an aqueous solution of sulfuric acid is formed.

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Darrell Ebbing has taught general chemistry for more than thirty years and is now retired from Wayne State University. He received his Ph.D. in physical chemistry from Indiana University. Steven D. Gammon is the Provost at the University of Maine Fort Kent.

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Each 1 H2 16 O molecule contains 8 neutrons, 10 protons, and 10 electrons. # neutrons in 6.0 1023 molecules = 8 (6.0 1023) = 4.8 1024 neutrons # protons in 6.0 1023 molecules = 10 (6.0 1023) = 6.0 1024 protons Because the water molecules are neutrally charged there must also be 6.0 1024 electrons. 2.146.